







WESTINGHOUSE LAMP CO

MANUFACTURERS OF

INCANDESCENT LAMPS

Moonfield, F. J. Sept. 20th, 1937.

Mr. Valter Macofile

Dear Mr. Russell's

The analysts of gas exepte #5 was reported

by the Chemical Laboratory as follows:

Oxygen 14.9

Hydrogen 16.0

Mitrogen er ::... 69,1 Inert gases

Yours very truly, WESTINCHOUSE LAWP COMPANY.

R. D. Hall, Sage.,

RDH:KW

This is one of 17 different results obtained from five cc of water in a sealed quartz tube with electroded on end for spectrum adalysis. In every case the resultant gases differed

On lighting match, force of water and gas coming out blew out the match and the pressure dropped from 45 to 40 pounds by the time Russell turned off valve. Hoyt lit another match and held above the valve instead of opposite, as he did before, so it wouldn't blow out.

There was a distinctive explosive noise as the hydrogen ignited. Then water vapor checked the orifice. Hoyt lighted another match and repeated the same thing till the flame died out with a pressure of 5 pounds.

A Hope

NOTES ON PRESSURE ON EXPERIMENT OF TUESDAY, AUG. 2, 1927.

When the water was pumped into the evacuated container, the monometer jumped from 3 mm. to about 70 mm which was lowered by further pumping to about 15 mm.

When the iron was red the air pressure gage on the container slowly moved to 16 pounds.

We then transferred it to the Transmutor losing 2 bounds on the way. Immediately after the container was in place and current turned on the pressure slowly rose to 45 pounds during a period of 8 minutes of cooling.

In Hah

At	10 -	16	pressure	gage	registered	35	pounds
At	10 -	17	tt.	Ħ	•	45	
At	10 -	18	11	Ħ	•	47	#

At 47 pounds opened valve and flame shot out two inches. It was intended to open valve very slightly and let gas out slowly but it opened suddenly so the gas shot out with violence igniting as it came out with a distinct explosive snap.

Hollon Rusull

Instead of the expensive and time consuming chemical method of obtaining free nitrogen in LIMITED quantities [Haber process], Nature's method would produce free nitrogen cheaply, quickly, and in UNLIMITED quantities. It is not necessary to call attention to the value to commerce and to agriculture, not to mention soil regeneration, that this method of obtaining nitrogen would be to the world.

In September, 1927, I demonstrated this principle of dual polarity control by arranging two pairs of solenoids - one pair with more windings than the other - in such a manner that the dual polarity of Nature was simulated.

With a steel or glass disc for an equator and a steel rod for amplitude, I adjusted my solenoids approximately to a plane angle where I roughly calculated oxygen belonged in its octave. I improvised an adjustment apparatus which would enable me to fasten any adjustment securely at any angle I chose.

I then inserted a few cubic centimeters of water in an evacuated quartz tube which had electrodes at each end for spectrum analysis readings.

Upon heating the tube in an electric furnace, and inserting it into the solenoid with the electric current turned on until the tube cooled, the first spectrum analysis showed over 80% to be hydrogen and the rest practically all helium. There was very little oxygen.

Each time I reset it, I obtained a new analysis. Whenever I set it so the north-south polarity was predominant because of using the stronger coils, the result gave more nitrogen. This was because the preponderant north-south polarity prolated the oxygen atom nucleus to its next higher tone.

When I reversed the polarity to east-west preponderance, the analysis showed more than its proper amount of oxygen and inert gases and less of hydrogen. This meant that preponderant east-west polarity had oblated the hydrogen nucleus.

The following analysis is a good example. When I took the tube to the laboratory, there was no water in it. That is why the analyst referred to his report as "gas sample No. 5," which follows:

Oxygen	14.9
Hydrogen	16.0
Nitrogen	69.1

It is needless to say that the above analysis shows east - west preponderance.

I am convinced that by proper adjustments mathematically worked out into formulas by experiment, free hydrogen, nitrogen or oxygen could be obtained without any trace of the others.

The only difference between the two methods of working [Haber Vs. Russll's transmutor] is that electricity is used as power in the laboratory without polarity control or gyroscopic guidance such as I made use of

When the gases have been sufficiently transformed by practice, the transformation of dense matter can then follow.